

# $\text{OH}^+$ and $\text{H}_2\text{O}^+$ : Probes of the Molecular Hydrogen Fraction and Cosmic-Ray Ionization Rate

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# Outline

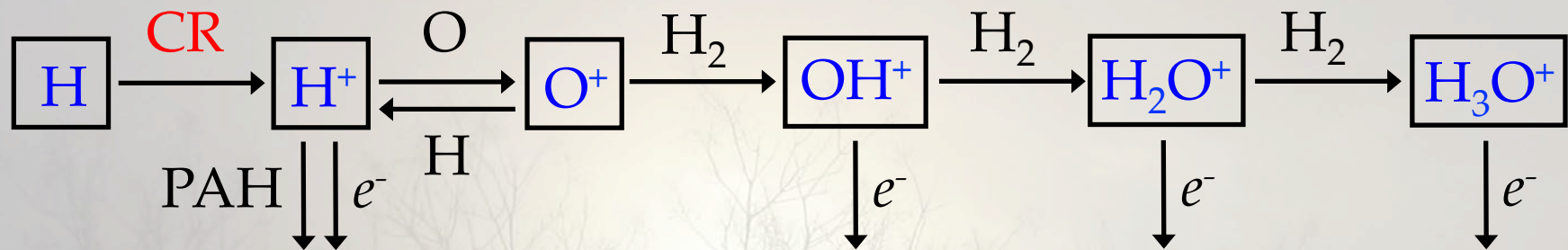
- Motivations
- Oxygen chemistry
- Observations
- Example Spectra
- Results
  - Molecular hydrogen fraction
  - Cosmic-ray ionization rate
- Where do we go from here?



# Motivation

- $\text{OH}^+$  and  $\text{H}_2\text{O}^+$  were unobserved prior to the launch *Herschel*
- Possible intermediaries in the process of  $\text{H}_2\text{O}$  formation
- Oxygen chemistry begins with ionization of atomic hydrogen, likely by cosmic rays
- Relative abundances of oxygen ions depend on competition between reactions with  $\text{H}_2$  and destruction via electrons

# Oxygen Chemistry



- $\text{CR} + \text{H} \rightarrow \text{H}^+ + e^- + \text{CR}'$
- $\text{H}^+ + \text{O} \rightarrow \text{O}^+ + \text{H}$
- $\text{O}^+ + \text{H}_2 \rightarrow \text{OH}^+ + \text{H}$
- $\text{OH}^+ + \text{H}_2 \rightarrow \text{H}_2\text{O}^+ + \text{H}$
- $\text{H}_2\text{O}^+ + \text{H}_2 \rightarrow \text{H}_3\text{O}^+ + \text{H}$

- $\text{OH}^+ + e^- \rightarrow \text{products}$
- $\text{H}_2\text{O}^+ + e^- \rightarrow \text{products}$
- $\text{H}_3\text{O}^+ + e^- \rightarrow \text{products}$
- $\text{O}^+ + \text{H} \rightarrow \text{H}^+ + \text{O}$
- $\text{H}^+ + e^- \rightarrow \text{H} + h\nu$
- $\text{H}^+ + \text{PAH} \rightarrow \text{PAH}^+$

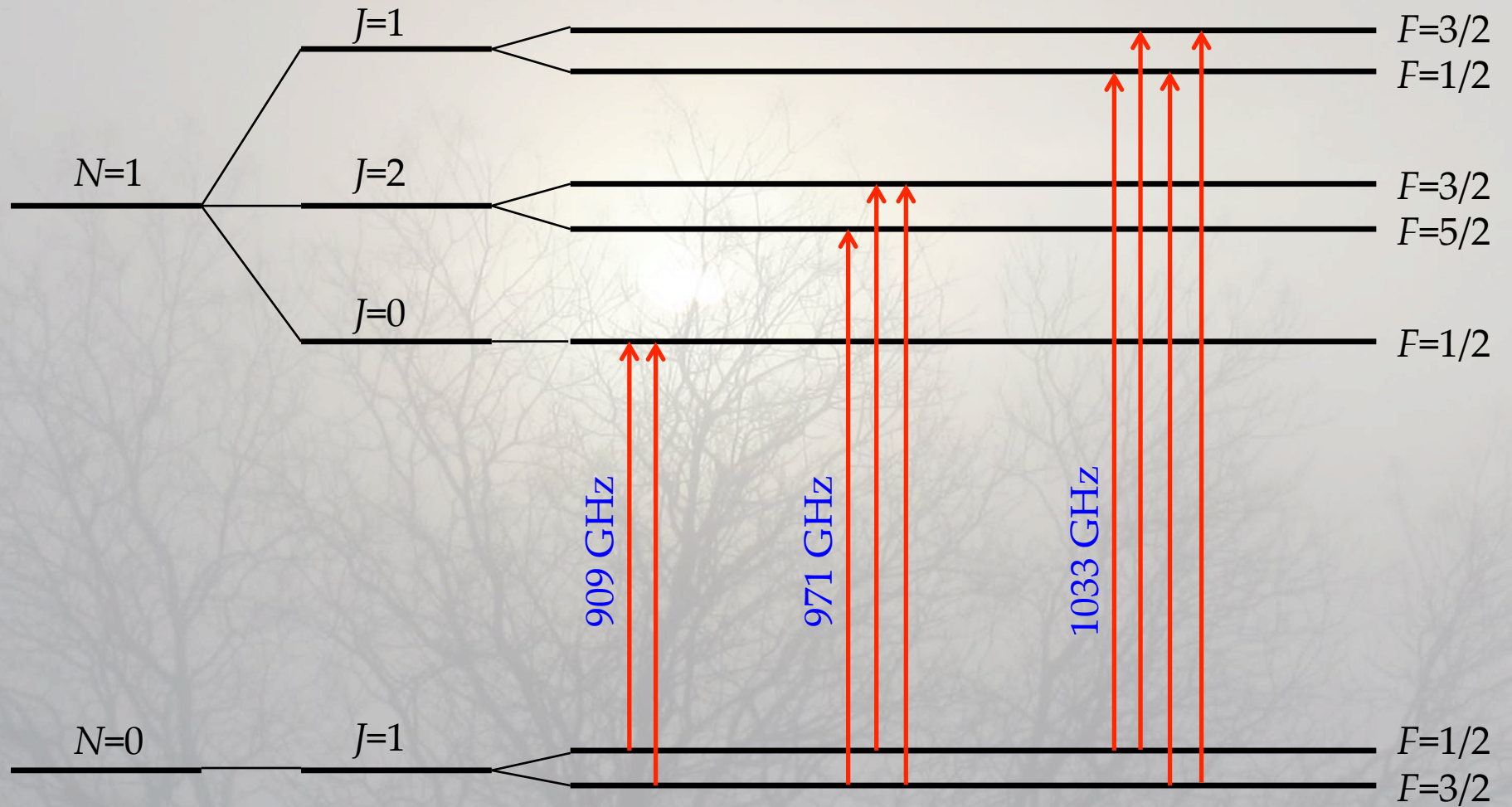


# Target List

Target	Right Ascension	Declination	Gal. Long.	Gal. Lat.	Program
SgrA* +20 km s <sup>-1</sup> cloud	17h 45m 37.4s	-29° 05' 40''0	359.8653	-0.0831	P
SgrA* +50 km s <sup>-1</sup> cloud	17h 45m 50.5s	-28° 59' 53''3	359.9724	-0.0737	P
W28A (G005.88-00.39)	18h 00m 30.5s	-24° 03' 59''9	5.8857	-0.3924	PW
W31C (G010.62-00.38)	18h 10m 28.7s	-19° 55' 50''0	10.6234	-0.3838	P
W33A (G012.91-00.26)	18h 14m 39.2s	-17° 52' 00''5	12.9078	-0.2592	PW
G029.96-00.02	18h 46m 03.8s	-02° 39' 22''0	29.9556	-0.0163	NW
G034.3+00.1 (W44C)	18h 53m 18.9s	+01° 14' 57''3	34.2577	0.1521	PW
W49N	19h 10m 13.1s	+09° 06' 12''5	43.1657	0.0123	P
W51 e <sub>1</sub> e <sub>2</sub> <sup>†</sup>	19h 23m 43.6s	+14° 30' 29''2	49.4879	-0.3871	NPW
AFGL 2591	20h 29m 24.7s	+40° 11' 18''7	78.8862	0.7091	W
DR21(OH)	20h 39m 00.9s	+42° 22' 48''6	81.7214	0.5713	PW
W3 IRS5	02h 25m 40.6s	+62° 05' 51''0	133.7168	1.2156	W
W3(OH)	02h 27m 03.8s	+61° 52' 24''6	133.9473	1.0642	N

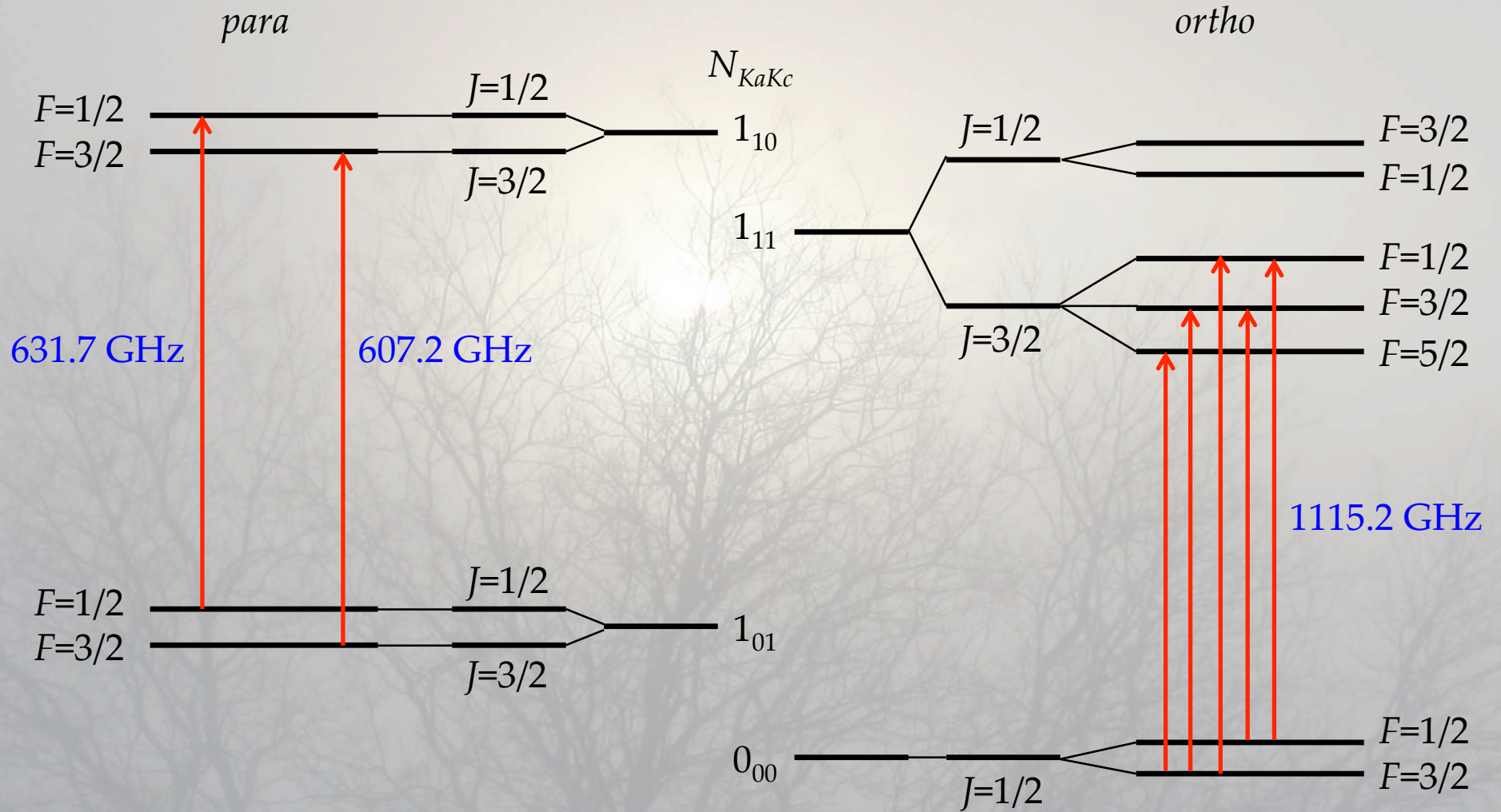
N – dneufeld\_1 (OT1)  
P – PRISMAS (KPGT)  
W – WISH (KPGT)

# OH<sup>+</sup> Transitions



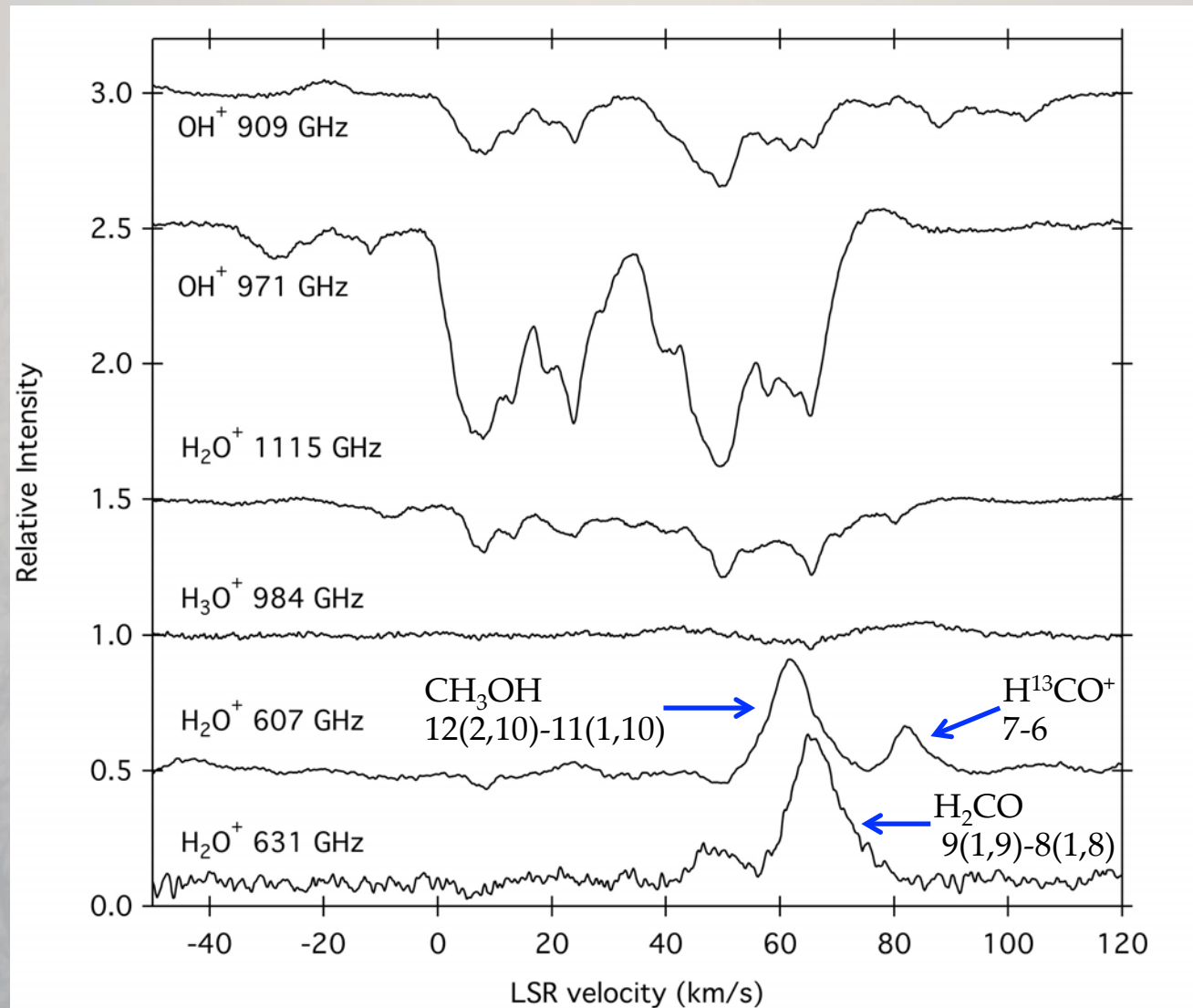


# H<sub>2</sub>O<sup>+</sup> Transitions

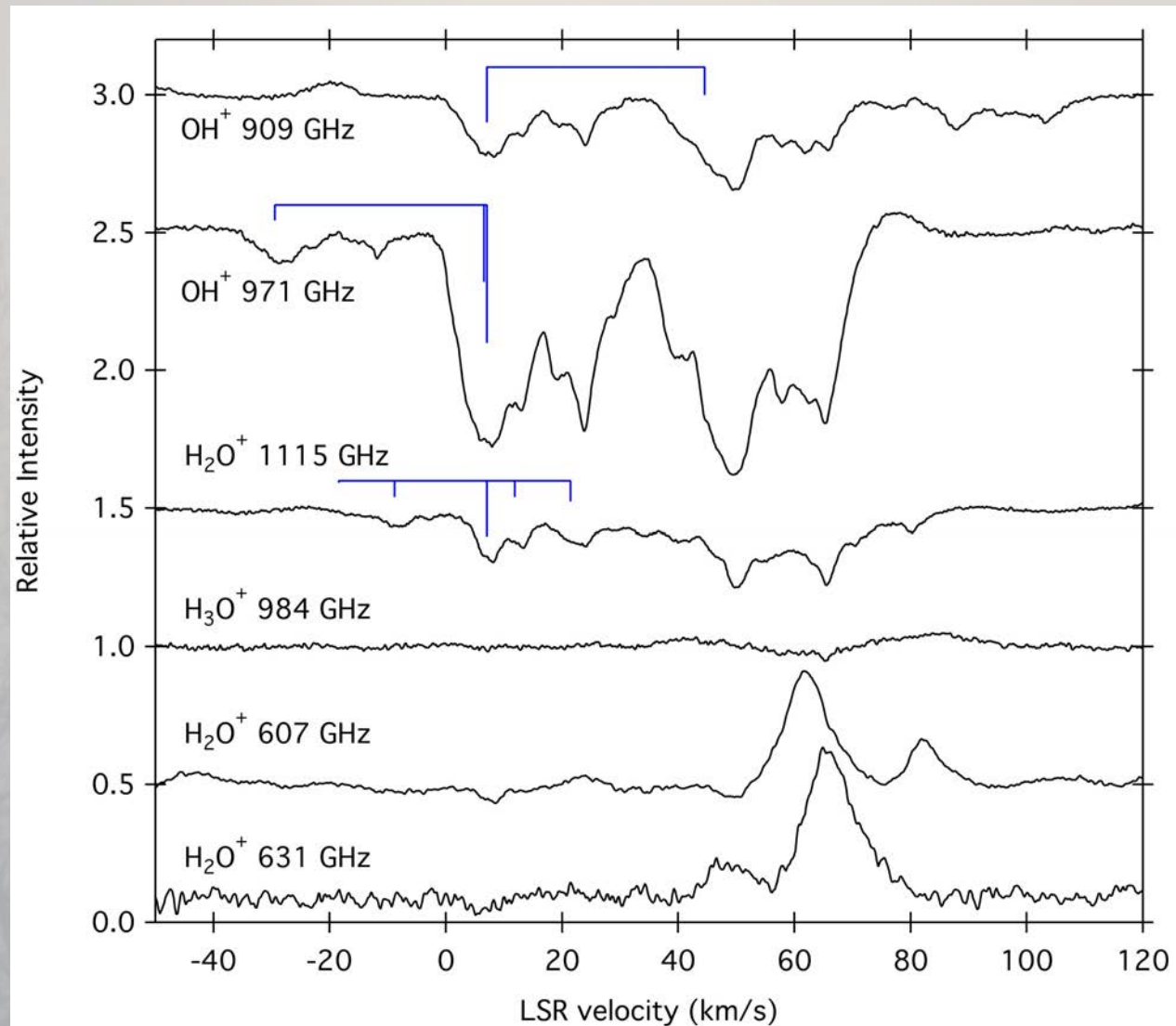




# W51 Spectra

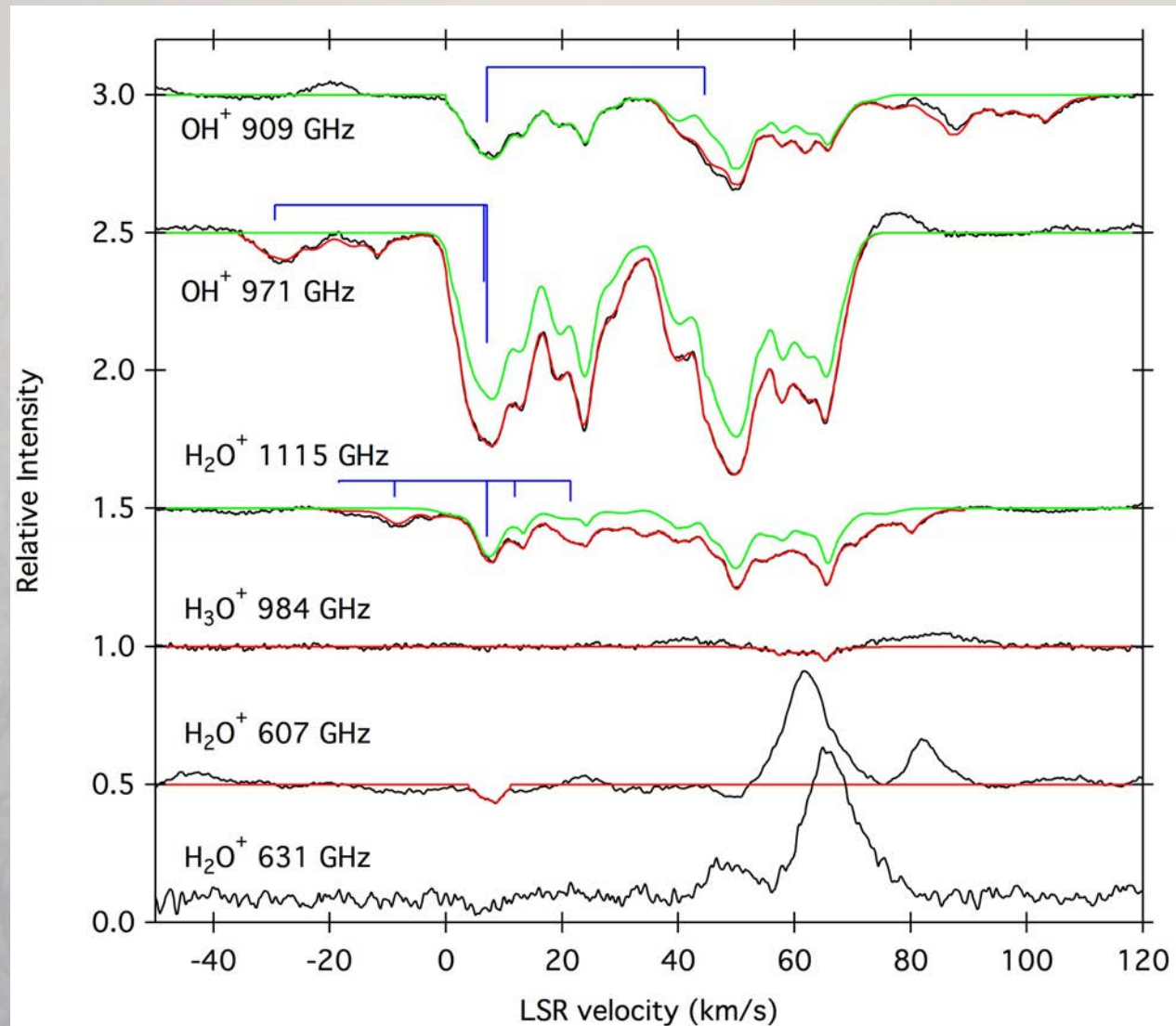


# W51 Spectra

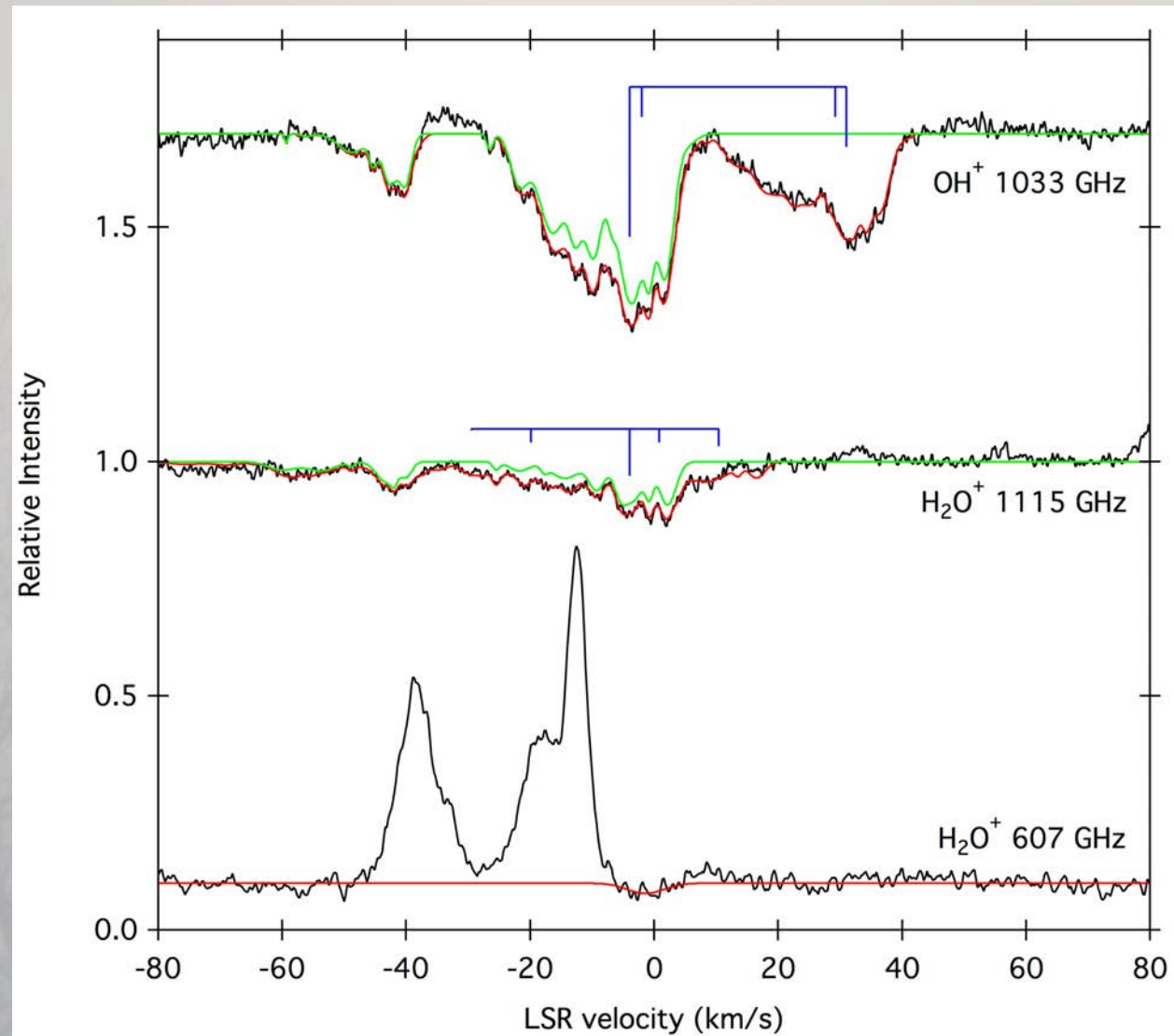




# W51 Spectra

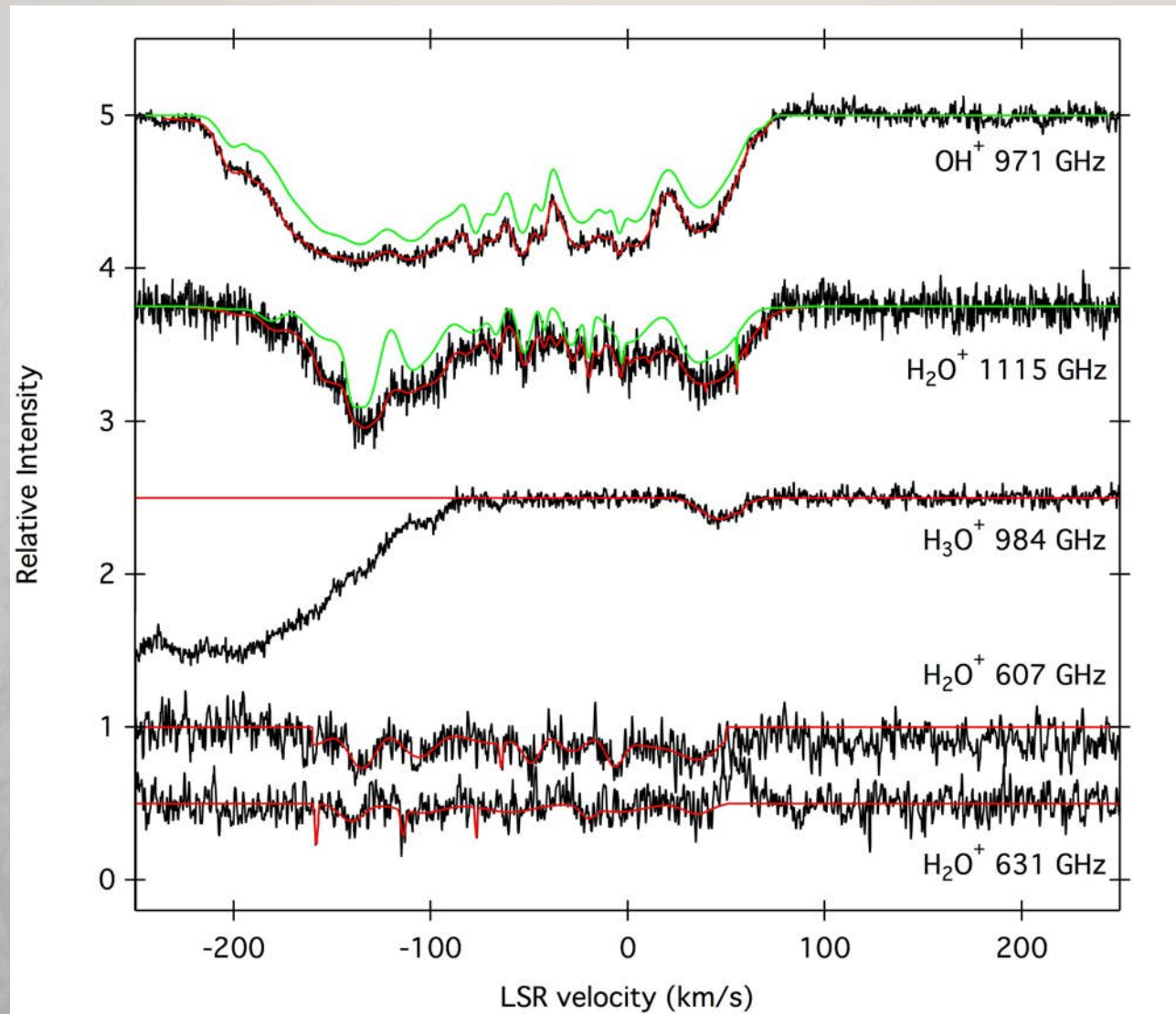


# W3 IRS 5 Spectra





# Sgr A\* +50 km/s Spectra



# W51 Results

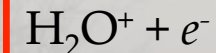
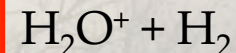
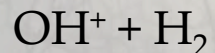
LSR velocity	$N(\text{OH}^+)$	$N(\text{o-H}_2\text{O}^+)$	$N(\text{p-H}_2\text{O}^+)$
(km/s)	( $10^{12} \text{ cm}^{-2}$ )	( $10^{12} \text{ cm}^{-2}$ )	( $10^{12} \text{ cm}^{-2}$ )
-4–11	32.9	4.52	1.53
11–16	11.1	1.34	<0.54
16–21	8.4	0.61	<0.51
21–33	18.5	1.35	<1.20
33–42	9.6	1.50	<0.93
42–55	52.2	7.45	
55–62	17.8	2.98	
62–75	20.4	4.73	



# Molecular Hydrogen Fraction

$$f(\text{H}_2) \equiv \frac{2n(\text{H}_2)}{n(\text{H}) + 2n(\text{H}_2)}$$

Assume steady-state chemistry for  $\text{H}_2\text{O}^+$



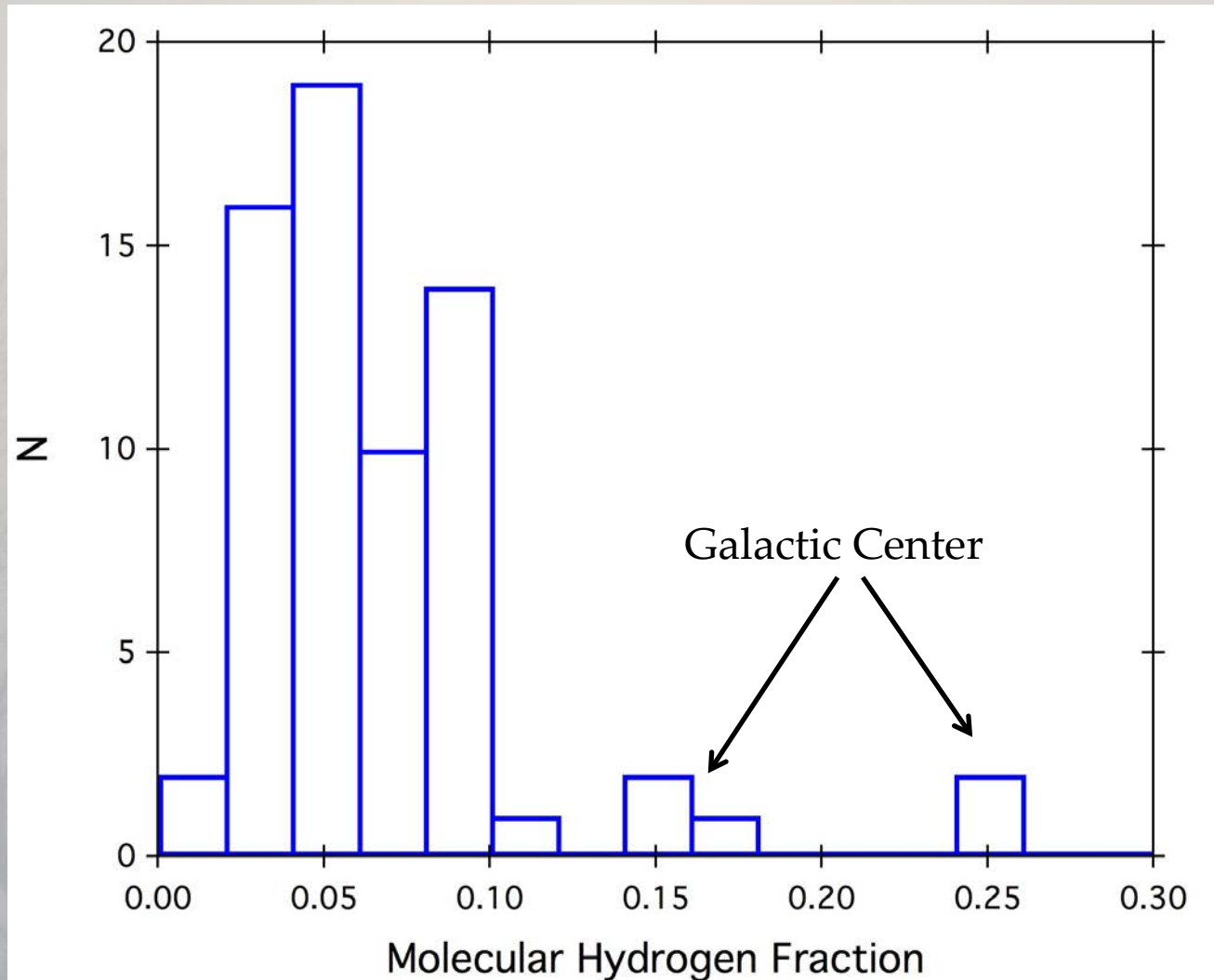
$$n(\text{OH}^+)n(\text{H}_2)k(\text{OH}^+|\text{H}_2) = n(\text{H}_2\text{O}^+)[n(\text{H}_2)k(\text{H}_2\text{O}^+|\text{H}_2) + n_e k(\text{H}_2\text{O}^+|e^-)]$$

$$f_{\text{H}_2} = \frac{2x_e k(\text{H}_2\text{O}^+|e^-)/k(\text{OH}^+|\text{H}_2)}{N(\text{OH}^+)/N(\text{H}_2\text{O}^+) - k(\text{H}_2\text{O}^+|\text{H}_2)/k(\text{OH}^+|\text{H}_2)}$$

# W51 Results Continued

LSR velocity	$N(\text{OH}^+)$	$N(\text{o-H}_2\text{O}^+)$	$N(\text{p-H}_2\text{O}^+)$	$f(\text{H}_2)$
(km/s)	( $10^{12} \text{ cm}^{-2}$ )	( $10^{12} \text{ cm}^{-2}$ )	( $10^{12} \text{ cm}^{-2}$ )	
-4–11	32.9	4.52	1.53	0.046
11–16	11.1	1.34	<0.54	0.040
16–21	8.4	0.61	<0.51	0.023
21–33	18.5	1.35	<1.20	0.023
33–42	9.6	1.50	<0.93	0.053
42–55	52.2	7.45		0.048
55–62	17.8	2.98		0.057
62–75	20.4	4.73		0.085

# Molecular Hydrogen Fraction





# Cosmic-Ray Ionization Rate

where  $\zeta_{\text{H}}$  is defined as the rate per H atom

Assume steady-state chemistry for  $\text{OH}^+$

CR + H

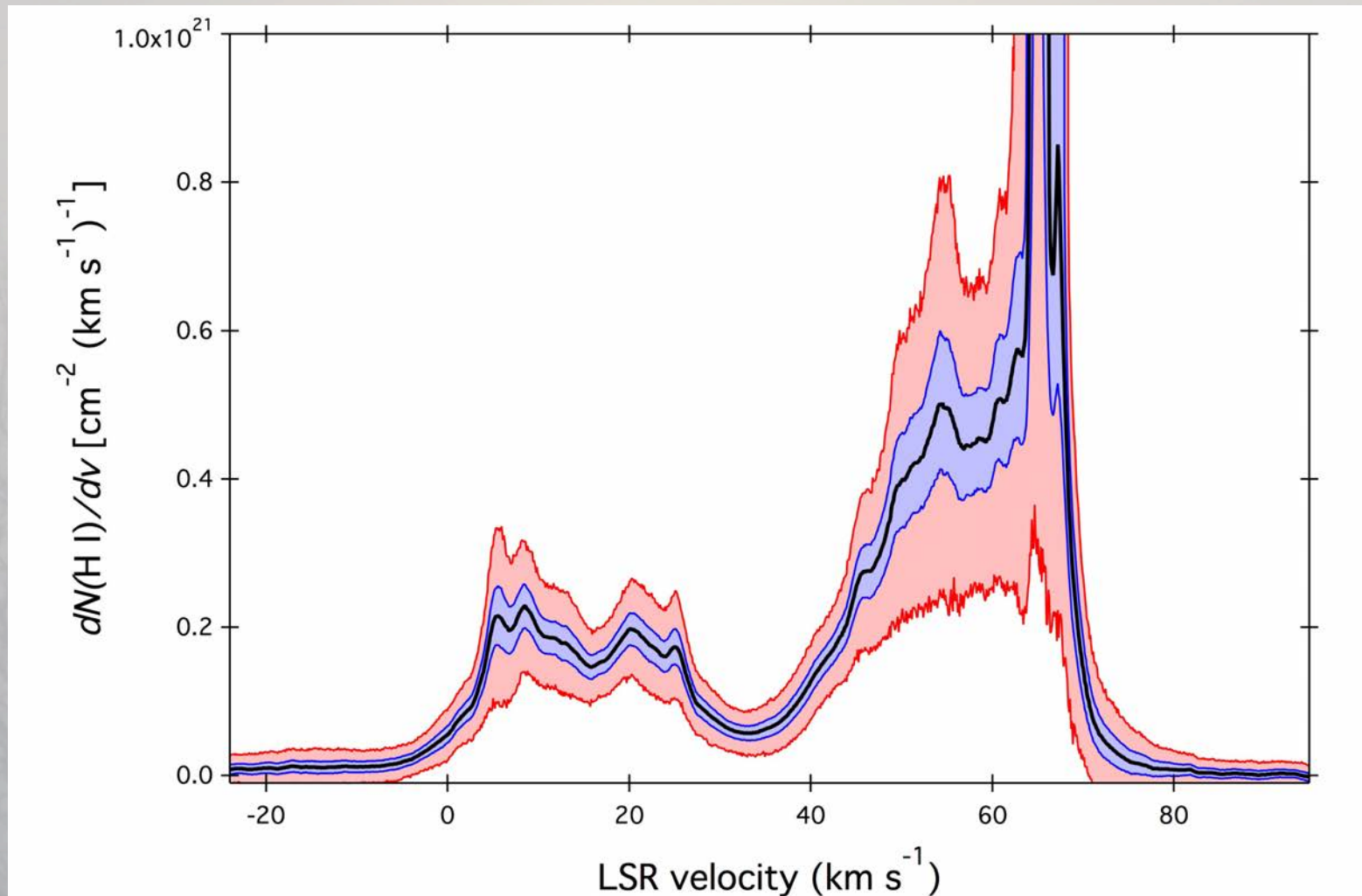
$\text{OH}^+ + \text{H}_2$

$\text{OH}^+ + e^-$

$$\epsilon \zeta_{\text{H}} n(\text{H}) = n(\text{OH}^+) [n(\text{H}_2) k(\text{OH}^+ | \text{H}_2) + n_e k(\text{OH}^+ | e^-)]$$

$$\epsilon \zeta_{\text{H}} = \frac{N(\text{OH}^+)}{N(\text{H})} n_{\text{H}} \left[ \frac{f_{\text{H}_2}}{2} k(\text{OH}^+ | \text{H}_2) + x_e k(\text{OH}^+ | e^-) \right]$$

# 21 cm Data for W51



H I data processed by Benjamin Winkel

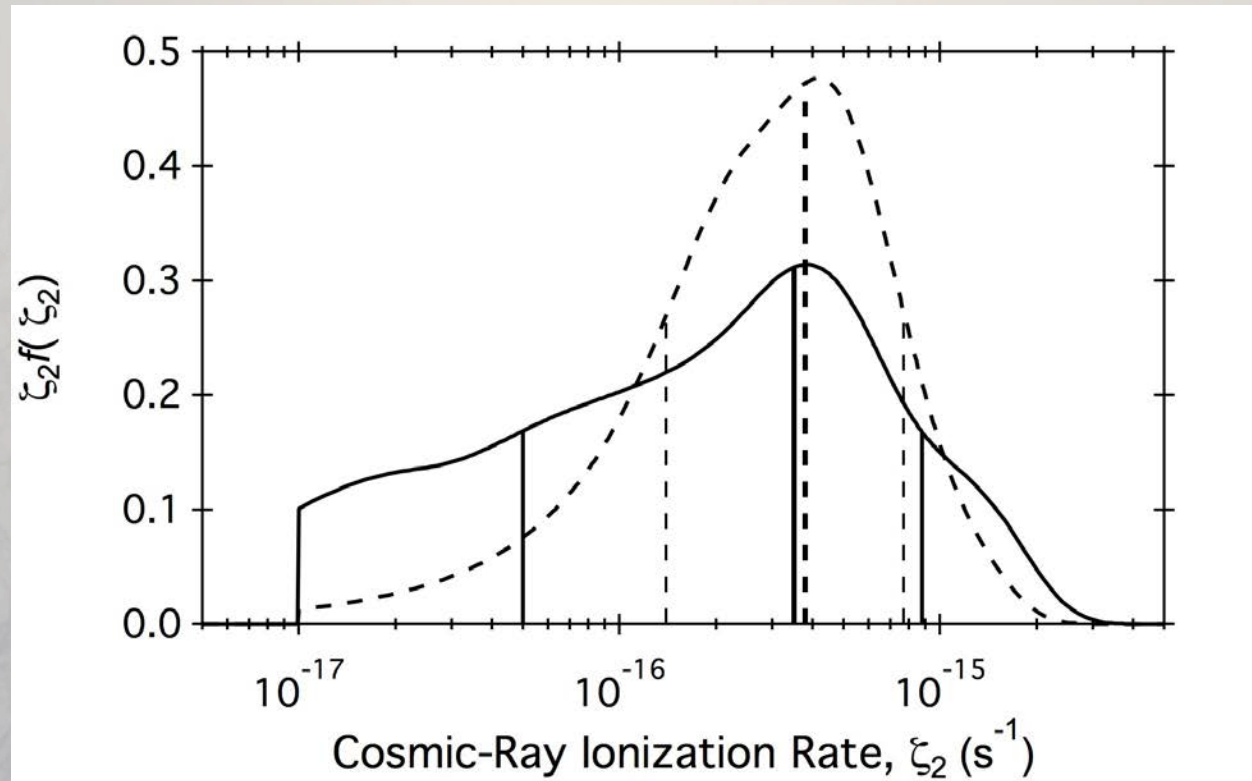
# W51 Results Concluded

LSR velocity	$N(\text{OH}^+)$	$N(\text{o-H}_2\text{O}^+)$	$N(\text{p-H}_2\text{O}^+)$	$f(\text{H}_2)$	$N(\text{H I})$	$\varepsilon\zeta_{\text{H}}$	$\zeta_2$
(km/s)	( $10^{12} \text{ cm}^{-2}$ )	( $10^{12} \text{ cm}^{-2}$ )	( $10^{12} \text{ cm}^{-2}$ )		( $10^{21} \text{ cm}^{-2}$ )	( $10^{-17} \text{ s}^{-1}$ )	( $10^{-16} \text{ s}^{-1}$ )
-4–11	32.9	4.52	1.53	0.046	1.94	1.96	4.29
11–16	11.1	1.34	<0.54	0.040	0.84	1.38	3.02
16–21	8.4	0.61	<0.51	0.023	0.88	0.71	1.56
21–33	18.5	1.35	<1.20	0.023	1.42	0.97	2.13
33–42	9.6	1.50	<0.93	0.053	0.84	1.46	3.21
42–55	52.2	7.45		0.048	4.43	1.40	3.06
55–62	17.8	2.98		0.057	3.25	0.74	1.63
62–75	20.4	4.73		0.085	>1.54	<2.44	<5.36

Assumed  $\varepsilon=0.07$



# Distribution of Ionization Rates



# Summary

- OH<sup>+</sup> and H<sub>2</sub>O<sup>+</sup> have been surveyed in 75 velocity components across 13 sight lines
- Column densities suggest these species reside in primarily atomic gas;  $f(\text{H}_2) < 0.1$
- The distribution of cosmic-ray ionization rates peaks near  $\zeta_2 = 4 \times 10^{-16} \text{ s}^{-1}$ , similar to results from H<sub>3</sub><sup>+</sup>, and is well probed down to about  $10^{-17} \text{ s}^{-1}$

# Future Work

- Search for correlations between the ionization rate and Galactocentric radius
- Consider alternative means of observing both  $\text{OH}^+$  (APEX, VLT UVES) and  $\text{H}_2\text{O}^+$  (VLT CRIRES)
- Combine results with those found by other tracers of the ionization rate to better understand the cosmic-ray flux and particle propagation