

(Sub)-millimeter line-surveys of the highmass star forming region G327.3-0.6

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HERSCHEL AND THE FORMATION OF STARS AND PLANETARY SYSTEMS

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- Introduction: Molecules in star forming regions
- Unbiased (sub)millimeter line-survey of the high-mass star-forming region G327.3-0.6

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$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccc} H_2 & & C_3 \\ AIF & & C_2H \\ AICI & & C_2O \\ C_2 & & C_2S \\ CH & & CH_2 \\ CH^+ & & HCN \\ CN & & HCO \\ CO & & HCO^+ \\ CO^+ & & HCS^+ \end{array}$	$c-C_3H$ C_5 $I-C_3H$ C_4H C_3N C_4Si C_3O $I-C_3H_2$ C_3S $c-C_3H_2$ C_2H_2 CH_2CN CH_2D^+ CH_4 HCCN HC_3N	$C_{5}H$ $I-H_{2}C_{4}$ $C_{2}H_{4}$ $CH_{3}CN$ $CH_{3}NC$ $CH_{3}OH$ $CH_{3}SH$ $HC_{3}NH^{+}$	$C_{6}H$ $CH_{2}CHCN$ $CH_{3}C_{2}H$ $HC_{5}N$ $HCOCH_{3}$ $NH_{2}CH_{3}$ $c-C_{2}H_{4}O$ $CH_{2}CHOH$	CH ₃ C ₃ N HCOOCH ₃ CH ₃ COOH C ₇ H CH ₂ OHCHO	$\begin{array}{c c} CH_{3}C_{4}H \\ CH_{3}CH_{2}CN \\ (CH_{3})_{2}O \\ CH_{3}CH_{2}OH \\ HC_{7}N \\ C_{8}H \\ CH_{3}C_{5}N \\ (CH_{3})_{2}CO \\ NH_{2}CH_{2}COOH? \\ HC_{0}N HC_{4}N \end{array}$
HCI H_2S HOCO+ H_2C_2O H_2C_2O NHHNC H_2CO H_2NCN NOHNO H_2CN NSMgCN H_2CS NaClMgNC H_3O^+ OH N_2H^+ NH_3PN N_2O SO^+OCSSiNSO_2SiO $c-SiC_2$	HCI H_2S KCI H_2S NH HNC NO HNO NOMgCNNSMgNCOH N_2H^+ PN N_2O SONaCNSO^+OCSSiN SO_2 SiOc-SiC ₂	HOCO ⁺ H_2CO H_2CO H_2CN H_2CS H_3O^+ NH_3 SiC_3 HOCO ⁺ H_2C_2O H_2NCN HNC_3 SiH_4 H_2COH	+ C ₅ N			

Overview chemistry during star formation

- Low temperatures: hydrogenation reactions of CO, O, N & C
- Intermediate temp: radical reactions in the ices
- High temp: gas phase reactions after evaporation



Chemistry useful a diagnostic tool but also interesting in itself: How complex do molecules become? Do/Can they end up in new planetary systems?

Figure 14

Cartoon representation of the evolution of material from the prestellar core stage through the collapsing envelope (size ~ 0.05 pc) into a protoplanetary disk. The formation of zeroth- and first-generation organic molecules in the ices is indicated with 0 and 1, and the second-generation molecules in the hot-core/corino region when the envelope temperature reaches 100 K, and even strongly bound ices start to evaporate, are designated 2. The grains are typically 0.1 μ m and are not drawn to scale. The temperature and density scale refer to the envelope, not to the disk (see also **Figure 4**). Once material enters the disk, it will rapidly move to the cold midplane where additional freeze-out and grain surface chemistry occur. All ices evaporate inside the (species-dependent) sublimation radius. For H₂O and trapped complex organic molecules, this "snow line" lies around a few astronomical units in a disk around a solar mass star. Figure by E. van Dishoeck & R. Visser.

Grain H₂O-rich

CO-ricl

Herbst & van Dishoeck, ARAA, 2009, 47, 427

Aim

Study the molecular composition of the surrounding material during the star formation process

Unbiased line-surveys:

- Large frequency coverage
- Low RMS
- Transitions of many different species (and isotopologues)

 Many transitions per species: excitation temperature, abundances, source sizes

The G327.3-0.6 star-forming region

- One of the chemically richest high-mass starforming regions known (Gibb et al. 2000)
- Strong and narrow emission lines (~ 5 km s⁻¹: minimal line-confusion)



Gibb et al., ApJ, 2000, 545, 309

G327.3-0.6 continued....

• Luminosity: 0.5-1.5 x10⁵ L •

• Distance: 2.9 kpc

• Mass: $\sim 500 \mathrm{M}_{\odot}$



Wyrowski et al., A&A, 2008, 454, L91 Grayscale: ¹²CO, contours: C¹⁸O (APEX)

Observations with the Atacama Pathfinder EXperiment (APEX)

APEX 1: 213 - 267.5 GHz

APEX 2: 270 - 315 & 335 - 362 GHz

CHAMP+: 623 -715 & 784 - 853 GHz



Data analysis

- Line-assignements: CDMS, JPL
- MyXCLASS package (Comito et al. 2005)
- Most molecules: 1/2 component fit, source size, T, column density, line width: hot core species ~2" with T~100-150 K
- 44 species detected, 52 isotopologues, 23 vibrationally excited species (e.g. HCN,HCO⁺ to (CH₃)₂CO)
- ~60-70% of lines assigned \Rightarrow 30+% U-lines

The incredible richness of G327.3-0.6



Overview detected species

1 atom: C

- 2-atoms: CN, SiO, SO, CS, NO, NS, CO
- 3-atoms: HCN, HNC, SO₂, H₂S, OCS, CCH, HCO⁺, N₂H⁺, HCS⁺, HDO/H₂¹⁸O, NH₂

4-atoms: HNCO, NH₂D, H₂CO, H₂CS

5-atoms: HC₃N, CH₂NH, H₂CCO, HCOOH, c-C₃H₂, l- C₃H₂

6-atoms: CH₃CN, CH₃NC, HCONH₂, CH₃SH, CH₃OH

Continued....

7-atoms: C_2H_3CN , CH_3CHO , CH_3NH_2 , $c-C_2H_4O$, CH_3CCH

8-atoms: HCOOCH₃

9-atoms: C_2H_5CN , CH_3OCH_3 , C_2H_5OH

10-atoms: CH₃CH₃CO

Many hydrogen-rich organics indicative of hydrogenation processes on grain-surfaces!

Case study I: CH₃OCH₃

- CH₃OCH₃ is formed on the surfaces of icy grains at higher temperature 20-30 K
- It evaporates ~70 K and destroyed in the gas phase
- As CH₃OH evaporates it is formed in the gas phase

Is there observational evidence for the grain surface / gas phase formation scenario?



Garrod et al., ApJ, 2008, 682, 283

CH₃OCH₃ in G327.3-0.6

- Ground state:
 - * 251 unblended lines detected over whole freq range
- Torsionally excited (new lab data by Endres et al. in prep):
 - ★ v11=1 (288 K, infrared inactive): 102 unblended lines
 - ★ v15=1 (346 K, infrared active): 43 unblended lines

CH₃OCH₃ continued



CH₃OCH₃ is detected in all atmospheric windows observed with transition spanning a large range in excitation energies

Excitation of CH₃OCH₃



- Comp 1: E_u> 200 K can be very well fit with a simple model with T~100 K and N~4.0x10¹⁸ cm⁻² and a source size of 2.3" (LTE)
- Comp 2: E_u< 200 K come from a region with a temperature of ~ 60 K and N~4.5x10¹⁶ cm⁻² for 4.0" (LTE??)

$CH_3OCH_3v_{11}=1$ and $v_{15}=1$ detected for the first time!

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 $C_H CN v =$

292.10

59.25

821.80

821.75

\$59.26

668.45



Recent lab. measurements by Endres et al. in prep. • $v_{11} = 1$ detected in all bands • $v_{15}=1$ detected at all but the highest frequencies Excitation well

described by identical model to the ground state

Summary CH₃OCH₃

• Main component of the ground state is tracing emission that has a temperature of 100 K

• A cooler more extended component is also present

Two formation mechanisms/ (non)-LTE excitation effects? Future work: modeling the CH₃OCH₃ with a more realisitic source model

Case study II: C₂H₅CN in G327.3-0.6

- Ground state:
 - ★ 324 unblended lines detected over whole freq range
 - Vibrationally / torsionally excited $v_{13}=1(298 \text{ K}, \text{CCN in-plane}) \& v_{21}=1 (303 \text{ K}, \text{torsion})$:

★ 237 detected lines at low frequencies (no laboratory data above 500 GHz)

Previously detected toward SgrB2(N) and W51 e2 (Mehringer et al. 2004, Demyk et al 2008)

Example II: C₂H₅CN



- v=0:
- Consistent with a single temperature and column density over a large excitation range
- Optically thick lines of all energies appear to probe the same source size of 2.3"
- ¹³C isotopes are also detected and fit the same model

Vibrationally and torsionally excited C₂H₅CN, v₁₃=1 and v₂₁=1



- Lines stronger than expected based on the ground state fit
- Some pumping mechanism?

Summary C₂H₅CN

The ground state is tracing emission with ~125 K, consistent with a grain evaporation origin of the gas

The column densities are within error the same for ground and vibrational states, but the excitation temperatures are not!

Does the vib/torsionally excited state arise from the same region as the ground state?

Future work: compare with related species such as C₂H₃CN!

General conclusion

 Detailed study of the excitation of complex organics in a line-survey gives a plethora of information of the whereabouts and possible formation mechanism of species

Interpretation of line-surveys is tricky and not straightforward!

Future line-surveys with Herschel and ALMA will serve as excellent to test our understanding of astrochemical processes!